# The Structures of Three Novel Sesquiterpenoids from Valeriana Jatamansi Jones 

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#### Abstract

Three new sesquiterpenoids valeriananoids A(1), B (2) and C (3), possessing an unprecedented sesquiterpene skeleton, have been isolated from the roots and mizomes of Voleriana jatamansi Jones. The structures were elucidated by chemical and spectroscopic means and, in the case of 1, confirmed by X-ray crystallography. © 1997 Elsevier Science Ltd.


Valeriana jatamansi Jones, distributed widely in southwestern areas of China, has been used as a traditional Chinese medicine due to its hypnotic, tranquilizing and antiviral activities. In our previous paper ${ }^{1}$ 40 chemical constituents of essentials oils from four Valeriana species have been reported. Described herein is the structure elucidation of three new sesquiterpenoids valeriananoids A-C (1-3) obtained from an EDOAc extract of the roots and rhizomes of Valeriana jatamansi Jones. Compounds 1-3 possess a unique carbon skeleton which has not been found in the sesquiterpenoids before. Valeriananoids B exhibited moderate in vitro antirotavirus activities .

$2 \mathbf{R = H}$
$3 \mathrm{R}=\mathrm{OAC}$


Figure 1. X-ray structure of 1

Compound 1 is a colorless crystal ( $\mathrm{Me}_{2} \mathrm{CO}$ ), m.p. $122-124{ }^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}{ }^{28}-34.6(\mathrm{c}=0.231, \mathrm{CHCl} 3)$. The molecular formula was established as $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{2}$ by EIMS ${ }^{2}$ and elemental analysis. IR absorptions at 3440 and $1710 \mathrm{~cm}^{-1}$ implied that 1 possessed both hydroxyl and carbonyl groups. Analyses of the ${ }^{1} \mathrm{H} N \mathrm{NR}$, ${ }^{13} \mathrm{C}$ NMR (Table 1 ) and DEPT data of 1 revealed the existence of one carbonyl, four methyls, four methylenes, three methines and three quaternary (one of them bearing an oxygen atom ) carbons. One out of the four unsaturations was thus characterized, and compound 1 was therefore inferred to contain three rings.
$\mathrm{H}-1 \mathrm{a}\left(\delta 1.77, \mathrm{~d} \mathrm{~J}=19.3\right.$ ) and $\mathrm{H}-1 \mathrm{~b}\left(\delta 2.63, \mathrm{~d} \mathrm{~J}=19.3\right.$ ) formed an AB system in the ${ }^{1} \mathrm{H}$ NMR spectrum of compound 1. The assignments of protons and carbons were obtained by the analysis of its 2D NMR spectra. The structure of compound 1 was confirmed by single crystal X- ray analysis. The compound 1 crystallized in the orthorhombic space group $\mathrm{P}_{1} 2121$ with one molecule of composition $\mathrm{C} 15 \mathrm{H} 24 \mathrm{O} 2(\mathrm{Z}=4)$ forming the asymmetric unit. Accurate cell constants of $a=10.133(2) \AA, b=11.566(3) \AA, c=11.600(3) \AA, v$ $=1359.62(8) \AA^{3}$ were determined by a least-squares fit of fifteen moderate $2 \theta$-angle values. All unique diffraction maxima with $2 \theta<114$ were collected on a Nicolet R3M/E using graphite monochromated $\mathrm{CuK} \alpha$ radiation ( $1.54178 \AA$ ) and variable speed $\omega$ scans. After correction for Lorentz, polarization and background effects, 1036 of the 1082 independent reflections were judged observed [ $I>3$ o(I)]. The structure was solved by SAPI programs and MAC science crystanG. Hydrogen atoms were located in a difference electron density synthesis after full least - squares refinement of the non-hydrogenatoms. The final conventional crystallographic residual is $\mathrm{R}=0.087$ for 1036 observed reflections. The final X -ray model is shown in figure 1. The final atomic parameters, bond lengths and angles will be deposited at the Cambridge Crystallographic Data Center. An unequivocal assignment of the absolute configuration was not achieved by X -ray diffraction . However, compound 1 showed positive Cotton effect at $300 \mathrm{~nm}(\triangle \varepsilon+0.21)$ in its $C D$ spectrum. This band could be assigned to the $n \rightarrow \pi *$ transition of the twisted cyclohexanone carbonyl group of $\mathbf{1}$. Application of the octant rules for twisted bridged-ring ketones ${ }^{5}$ suggested for this compound the absolute configuration shown in the structure 1 .

Compound 2 , colorless needles, m.p. 200-202 ${ }^{\circ} \mathrm{C}$, $[\alpha]_{D^{28}}-57.87$ ( $\mathrm{c}=0.197, \mathrm{CHCl}_{3}$ ), was determined to have the molecular formula $\mathrm{C}_{1} \mathrm{SH}_{26} \mathrm{O}_{2}$ by EIMS ${ }^{3}$ and elemental analysis. IR absorptions ${ }^{3}$ at 3420,3380 $\mathrm{cm}^{-1}$ suggested the presence of hydroxy. ${ }^{1} \mathrm{H},{ }^{13} \mathrm{C}$ NMR data (Table 1) and DEPT spectra of 2 revealed 15 signals: four methyls, four methylenes, four methines (including an oxymethine), three quaternary (one of them bearing an oxygen atom ) carbons. Comparison of spectroscopic data of the compound 1 with those of compound 2 showed that the carbonyl group of 1 was replaced in 2 by a hydroxyl group. Compound 3 was also obtained as colorless needles ( $\mathrm{Me}_{2} \mathrm{CO}$ ), m.p. $79-80^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}{ }^{28}-56.88$ ( $c=0.160 \mathrm{CHCl} 3$ ). Its IR spectrum ${ }^{4}$ contained absorptions for hydroxyl $\left(3505 \mathrm{~cm}^{-1}\right)$ and ester $\left(1715 \mathrm{~cm}^{-1}\right)$ functional groups and some structural features found in 1 and 2. The identification of 17 carbons in the ${ }^{13} \mathrm{C}$ NMR (Table 1) and DEPT experiment, and the EIMS ${ }^{4}$ indicated a molecular formula of $\mathrm{C}_{17} \mathrm{H}_{28} \mathrm{O}_{3}$. Compound 1 and 3 showed very similar NMR spectra except for that the signals of one acetyl group were present only in the spectra of 3. Moreover, hydrolysis ( $\mathrm{KOH} / 95 \% \mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}$ ) of 3 afforded 2 and oxidation $\left(\mathrm{CrO}_{3} / \mathrm{py}\right)$ of 2 gave 1
(scheme 1). Both 2 and 1 thus obtained were identified by comparison of their $R_{f}$ 's, mp's, IR and MS. Thus the structures of valeriananoids $B$ and $C$ were concluded to be 2 and 3 , respectively .


Scheme 1. Transformation of 3 into 2 and 1.

Table 1. NMR data for 1, 2 and 3.

| Compound | 1 |  | 2 |  | 3 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Position | ${ }^{1} \mathrm{H}$ | ${ }^{13} \mathrm{C}$ | ${ }^{1} \mathrm{H}$ | ${ }^{13} \mathrm{C}$ | ${ }^{1} \mathrm{H}$ | ${ }^{13} \mathrm{C}$ |
| la | $2.63 \mathrm{~d}(19.3)$ |  | $2.00 \mathrm{dd}(13.6,6.6)$ |  | 1.95dd(14.1, 7.6) | 36 |
| 1b | $1.77 \mathrm{~d}(19.3)$ | 46.5 | 1.57dd(13.6,7.8) | 39.5 | $1.63 \mathrm{dd}(14.1 .10 .4)$ |  |
| 2 |  | 215.9 | $3.84 \mathrm{ddd}(7.8,6.6,1.4)$ | 72.6 | 4.78ddd(10.4.7.6.1.8) | 74.6 |
| 3 | 1.84dd(5.0.5.6) | 57.7 | 1.44t(7.5) | 46.8 | 1.46t(6.4) | 43.4 |
| 4 |  | 40.5 |  | 39.5 |  | 39.1 |
| 5 |  | 74.3 |  | 77.2 |  | 75.0 |
| 6 a | 1.72m | 33.1 | 1.50 m | 32.6 | 1.51 m | 32.7 |
| 6 b | 1.88 m |  | 1.70 m |  | 1.74 m |  |
| 7 a | 1.41 m | 28.3 | 1.31 m | 28.6 | 1.34 m | 28.6 |
| 7 b | 1.58 m |  | 1.52 m |  | 1.53 m |  |
| 8 | 2.04 m | 27.9 | 1.93m | 27.7 | 1.94 m | 27.7 |
| 9 | 1.53 m | 43.6 | 1.37 m | 42.5 | 1.41 m | 42.5 |
| 10 |  | 42.4 |  | 40.9 |  | 40.1 |
| 11 a | 1.43 m | 20.7 | 1.35 m | 24.1 | 1.37 m | 23.5 |
| 11 b | 1.61 m |  | 1.53 m |  | 1.54 m |  |
| 12-Me | 0.84d(6.7) | 16.7 | 0.77 d (6.7) | 16.9 | 0.80d(6.7) | 18.7 |
| 13-Me | 1.17s | 23.0 | 1.20s | 25.4 | 1.14s | 25.0 |
| 14-Me | 1.01 s | 27.7 | 1.05s | 28.1 | 1.10 s | 27.8 |
| 15-Me | 0.97 | 19.4 | 0.84s | 20.2 | 0.89s | 20.0 |
| 16 |  |  |  |  |  | 170.8 |
| 17 |  |  |  |  | 2.01 s | 21.5 |

[^0]The ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ and ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ COSY NMR spectra ( $500 \mathrm{MHz}, \mathrm{CDCl} 3$ ) of 3 established the relationships between the signals at $\delta 0.80(\mathrm{Me}-12)$ and $1.94(\mathrm{H}-8) ; \delta 1.95,1.63(\mathrm{H}-1)$ and $4.78(\mathrm{H}-2) ; \delta 1.51,1.74(\mathrm{H}-6)$ and $1.34,1.53(\mathrm{H}-7) ; 81.94(\mathrm{H}-8)$ and $1.41(\mathrm{H}-9) ; 81.46(\mathrm{H}-3)$ and $1.37,1.54(\mathrm{H}-11)$, which allowed the unambiguous assignment of the upfield protons such as $\mathrm{H}-1 \mathrm{a}$ and $\mathrm{H}-8, \mathrm{H}-6$ and $\mathrm{H}-7$, and revealed connectivities of $\mathrm{C}-1$ to $\mathrm{C}-2, \mathrm{C}-6$ to $\mathrm{C}-7, \mathrm{C}-8$ to $\mathrm{C}-9, \mathrm{C}-9$ to $\mathrm{C}-11$, and $\mathrm{C}-11$ to $\mathrm{C}-3$. Of interest was the absence of coupling between $\mathrm{H}-2$ and $\mathrm{H}-3$, which suggested a torsion angle of nearly $90^{\circ}$ (it can be observed in Dreiding models). The $w$ type long-range coupling between $\mathrm{H}-2$ and $\mathrm{H}-11$ was observed in the ${ }^{1} \mathrm{H}$ NMR spectrum, along with the couplings between $\mathrm{H}-2$ and $\mathrm{H}-1 \mathrm{a}$ as well as $\mathrm{H}-1 \mathrm{~b}$. The results indicated a $\beta$-oriented acetoxy group at $\mathrm{C}-2$ since only $\alpha \mathrm{H}$ would produce $w$ type long- range coupling with H-11. The assignments of protons and carbons of $\mathbf{3}$ were summarized in Table 1.

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## References and Notes

1. Ming, Dongsheng; Guo, Jixian. Chinese Traditional Patent Medicine, 1994, 16 (1) , 41.
2. Spectroscopic data for valeriananoids $A(1)$ : ElMS $\mathrm{m} / \mathrm{z}$ (rel. intensity ) $\left[\mathrm{M}^{+}\right] 236(33), 201(5), 153(33)$, 135(33), 125(95), 109(25), 107 (20), $96(15), 83(100), 69(40), 55(42), 42(66)$. IR ( $\left.\mathrm{cm}^{-1}\right): 3440,1720$, $1470,1390,1340,1220,1100,1070,10501000,890,590$.
3. Spectroscopic data for valeriananoids B (2): EIMS m/z (rel. intensity) [ $\left.\mathrm{M}^{+}\right]$238(0.4), 220(5), 205(5), 159(4), 15(3), 125(100), 107(12), 96(45), 81(13), 55(15), 42(20). IR( $\left.\mathrm{cm}^{-1}\right): 3420,3380,1470$, $1380,1340,1170,1110,1050,1010,920,620$.
4. Spectroscopic data for valeriananoids C (3): EIMS m/z (rel. intensity) $\left[\mathrm{M}^{+}-\mathrm{AOOH}\right] 220(12), 205(13)$, 203(10), 177(12), 125(100), 109(9), 107(18), 96(80), 81(13), 69(5), 55(9), 42(40). IR (cm ${ }^{-1}$ ): 3500, $1720,1470,1400,1370,1270,1210,1160,1030,1010,980,900,560$.
5. Kirk, D.N. Tetrahedron, 1986, 42 (3), 777.
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[^0]:    * Spectra were recorded at 500 MHz for ${ }^{1} \mathrm{H}$ and 125 MHz for ${ }^{13} \mathrm{C}(\mathrm{CDCl} 3)$, with chemical shifts ( 8 ) in ppm and coupling constants in Hz put in ().

